Synthesis of a Potent Antagonist of E-Selectin

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A synthesis of an antagonist of E-selectin previously reported by a group at Novartis Pharma in Basel is described. An important feature involves the formation of an ether linkage based on a Rh^{II}-catalyzed reaction. Stereocontrolled glycosylations rely on the anomeric activation of 2-pyridylthio carbonate as leaving group for the attachment of β -D-galactopyranosyl and α -L-fucopyranosyl units on a common 1,5-anhydro D-glucitol scaffold.

Introduction

The recruitment and extravasation of leukocytes from the blood stream to the locus of inflammation in tissues is an essential immune-modulated process leading to beneficial effects and healing.¹ However, excessive recruitment and influx of leukocytes at the site of injury may result in adverse reactions that can be manifested in reperfusion injuries, stroke, rheumatoid arthritis, asthma, diabetes, and other acute or chronic health problems.² The phenomenological and molecular basis for initiating the physiological events associated with the over-recruitment of leukocytes are reasonably well understood.³ Endothelial cells lining the inner walls of blood vessels express adhesion molecules called E- and Pselectin when activated or induced. The sialyl Lewis^x tetrasaccharide expressed on the surface of glycoproteins of leukocytes are common epitopes recognized by the selectins.⁴ In a cascade of events, leukocytes are recognized by these physiological ligands through specific interactions with pharmacophoric groups on the sialyl Lewis^x epitope. A rolling process of leukocytes on the endothelial cells at the inner blood vessels ensues, leading to excessive recruitment at the site of tissue inflammation. What follows is extravasation into the affected sites with beneficial or adverse effects.

Extensive studies on the design and synthesis of molecules that mimic sially Lewis $\!\!^x$ have been reported

(2) See, for example: (a) Diaz-González, F.: Sánchez-Madrid, F.
 Immunol. Today 1998, 169. (b) Mousa, S. A.; Cheresh, D. A. *Drug Discovery Today* 1997, *2*, 187. (c) Springer, T. A. *Cell* 1994, *76*, 301.
 (d) Varki, A. *Curr. Opin. Cell Biol.* 1992, *4*, 257. (e) Musser, J. H. *Annu. Rev. Med. Chem.* 1992, *27*, 301 and references therein.

(3) For selected reviews and references, see: (a) Alon, R.; Hammer, D. A.; Springer, T. A. Nature 1995, 374, 539. (b) Springer, T. A. Cell 1994, 76, 301. (c) Lasky, L. A. Science 1992, 258, 964. (d) Hammer, D. A.; Apte, S. M. Biophys. J. 1992, 63, 35. (e) Osborn, L. Cell 1990, 62, 3. (f) Carlos, T. M.; Harlan, J. M. Immunol. Rev. 1990, 114, 5. (g) Stoolman, L. M. Cell 1989, 56, 907 and references therein. (4) (a) Phillips, M. L.; Nudelman, E.; Gaeta, F. C. A.; Perez, M.;

(4) (a) Phillips, M. L.; Nudelman, E.; Gaeta, F. C. A.; Perez, M.; Sinhgal, A. K.; Hakomori, S.-I.; Paulson, J. C. *Science* **1990**, *250*, 1130.
(b) Walz, G.; Aruffo, A.; Kolanus, W.; Bevilacqua, M. D.; Seed, B. *Science* **1990**, *250*, 1132. (e) Berg, E. L.; Robinson, M. K.; Mansson, O.; Butcher, E. C.; Magnani, J. L. J. Biol. Chem. **1991**, *266*, 14869. in an effort to develop antagonists to the natural receptors such as E-selectin.⁵ Numerous carbohydrate-type⁶ and related structures consisting of hybrid molecules⁷ have been reported, with binding affinities in the low micromolar level.⁸ A common design element has capitalized on the nature of the known pharmacophores involved in the recognition between sialyl Lewis^x and E-selectin. Thus, the 3,4-hydroxyl groups of L-fucosyl and the 6-hydroxyl group of D-galactosyl units are essential for

M. B., Levy, D. E. Chill, Fhalm. Des. **1993**, 1, 221.
(6) For selected examples, see ref 5; see also: (a) Ramphal, J. Y.; Hiroshige, M.; Lou, B.; Gaudino, J. J.; Hayashi, M.; Chen, S. M.; Chiang, L. C.; Gaeta, F. C. A.; De Frees, S. A. J. Med. Chem. **1996**, 39, 1357. (b) Wong, C.-H.; Halcomb, R. L.; Ichikawa, Y.; Kajimoto, T. Angew. Chem., Int. Ed. Engl. **1995**, 34, 505. (c) Ohmoto, H.; Kakamura, K.; Inoue, T.; Kondo, H.; Inone, Y.; Yoshino, K.; Kondo, H.; Kakamura, K.; Inoue, T.; Kondo, H.; Inone, Y.; Yoshino, K.; Kondo, H.; Kakamura, K.; Inoue, T.; Kondo, H.; Inone, Y.; Yoshino, K.; Kondo, H.; Kakamura, K.; Inoue, T.; Kondo, J.; Bertozzi, D. R.; Pohl, N. L.; Rosen, S. D.; Kiessling, L. L. J. Org. Chem. **1995**, 60, 6254.

7) For peptide and amino acid-based analogues, see: (a) Hanessian, S.; Huynh, H. K.; Reddy, G. V.; McNaughton-Smith, G.; Ernst, B.; Kolb, H. C.; Magnani J. L. Bioorg. Med. Chem. Lett. 1998, 8, 2803. (b) Tsukida, T.; Moriyama, H.; Kurokawa, K.; Achiha, T.; Inoue, Y.; Kondo, H. J. Med. Chem. **1998**, 41, 4279. (c) Kurokawa, K.; Kumihara, H.; Kondo, H. Bioorg. Med. Chem. Lett. **2000**, 10, 1827. (d) Tsai, C.-Y.; Park, W. K. C.; Weitz-Schmidt, G.; Ernst, B.; Wong, C.-H. Bioorg. Med. *Chem. Lett.* **1998**, *8*, 2333. For analogues with partial carbohydrate components, see: (e) Kaila, N.; Thomas, B. E., IV; Thakker, P.; Alvarez, J. C.; Camphausen, R.; Crommie, D. Bioorg. Med. Chem. Lett. 2001, 10, 151. (f) Kretzchmar, G. Tetrahedron 1998, 54, 3765. (g) Murphy, P. V.; Hubbard, R. E.; Manallack, D. T.; Montana, J. G.; Taylor, R. J K. Tetrahedron Lett. 1998, 39, 3273. (h) Hanessian, S.; Reddy, G. V. Huynh, H. K.; Pan, J.; Pedatella, S.; Ernst, B.; Kolb, H. C. Bioorg. Med. Chem. Lett. **1997**, *7*, 2729. (i) Bamford, M. J.; Bird, M.; Gore, P. M.; Holmes, D. S.; Priest, R.; Prodger, J. C.; Saez, V. Bioorg. Med. Chem. Lett. **1996**, *6*, 239. (j) Liu, A.; Dillon, K.; Campbell, R. M.; Cox, D.; Huryn, D. M. Tetrahedron Lett. **1996**, *37*, 3785. (k) Kaila, N.; Yu, H.-A.; Xiang, Y. Tetrahedron Lett. 1995, 36, 5503. (I) Heskamp, B. M.; Veeneman, G. H.; van der Marel, G. A.; van Boeckel, C. A. A.; van Boom, J. H. Rec. Trav. Chim. Pays-Bas 1995, 114, 398. (m) Kogan, T. P.; Dupré, B.; Keller, K. M.; Scott, I. L.; Bui, H.; Market, R. V.; Beck,
 P. J.; Voytus, J. A.; Revelle, B. M.; Scott, D. *J. Med. Chem.* 1995, *38*,
 4976. (n) Allanson, N. M.; Davidson, A. H.; Floyd, C. D.; Martin, F. M. Tetrahedron: Asymmetry 1994, 5, 2061. (o) Ragan, J. A.; Cooper, K. Bioorg. Med. Chem. Lett. 1994, 4, 2563 (p) Hanessian, S.; Prabhanjan, Bloorg. Med. Chem. Lett. 1994, 4, 2505 (p) Hancsslan, G, Habraham, H. Synlett 1994, 4, 2563. For an example of a non-carbohydrate analogue, see: (m) De Vleeschawer, M.; Vaillancourt, M.; Goudreau, N.; Guindon, Y.; Gravel, D. Bioorg. Med. Chem. Lett. 2001, 11, 1109.
(8) For comments on the binding affinities of E- and P-selectin the selection of the selection of the selection of the selection of the selection. antagonists, see: Kretschmar, G.; Toepfer, A.; Hüls, C.; Krause, M. Tetrahedron 1997, 53, 2485.

^{*} To whom correspondence should be addressed. Tel: (514) 343-6738. Fax: (514) 343-5728.

For pertinent reviews, see (a) Vestweber, D.; Blanks, J. E. Physiol. Rev. 1999, 79, 181. (b) Lasky, L. A. Annu. Rev. Biochem. 1995, 64, 113. (c) Bevilacqua, M. D. Annu. Rev. Immunol. 1993, 11, 767. (d) Paulson, J. C. In Adhesion, Its Role in Inflammatory Disease; Harlan, J. M., Liu, D. Y., Eds.; Freeman, New York, 1992; Chapter 2, p 19.

⁽⁵⁾ For pertinent reviews, see: (a) Simanek, E. E.; McGarvey, G. J.; Jablonski, J. A.; Wong, C.-H. *Chem. Rev.* **1998**, *98*, 833. (b) Roy, R. In *Carbohydrates in Drug Design*; Witczak, Z. J., Nieforth, K. A., Eds.; Marcel Dekker: New York, 1997; p 83. (d) Musser, J. H.; Anderson, M. B.; Levy, D. E. *Curr. Pharm. Des.* **1995**, *1*, 221.

binding to a calcium ion.⁹ The carboxylic acid group is involved in the formation of a salt bridge with an arginine residue in the receptor. Extensive studies involving molecular modeling,¹⁰ NMR,¹¹ and X-ray crystallography¹² have delineated the importance of attaining an optimal bioactive conformation for efficient binding to E-selectin. Within the group of carbohydrate-type pseudosaccharide antagonists, the incorporation of an α -Lfucosyl and a β -D-galactosyl unit that strategically mimic their counterparts in sialyl Lewis^x appears to be essential for good binding. The GlcNAc moiety can be replaced by simple carboxylic^{13,14} or heterocyclic units.^{7a} Perhaps the most revealing modification was the replacement of the N-acetylneuraminyl portion in sialyl Lewis^x with an S-cyclohexyl-2-propionic acid group attached as an α -ether linkage to the C-4 position of D-gal.^{15,16} A series of related analogues and their binding affinities expressed as IC₅₀ values is shown in Figure 1. Starting with sialyl Lewis^x 1 at $IC_{50} = 1000 \ \mu M$, it is clear that considerable flexibility exists with regard to the GlcNAc portion provided the replacement motif is cyclic. The need for a hydrophobic unit associated with the N-acetylneuraminyl portion is evident in comparing glycolic acid and α -substituted variants. The most active analogue is represented by structure 2,16 which is represented in two perspective drawings. This structure was arrived at by refinements in the GlcNAc mimetic portion by replacing it with an 1.5-anhvdro-2-deoxy-D-xvlo-hexitol unit.¹⁷ to which were glycosidically linked β -D-gal and α -L-fucosyl units at C-3 and C-4, respectively. The importance of

(10) (a) Tsujishita, H.; Hiramatsu, Y.; Kondo, H.; Kiso, M.; Hasegawa, A. J. Med. Chem. **1997**, 40, 362. (b) Coterón, J. M.; Singh, K.; Asencio, J. L.; Dominguez-Dalda, M.; Fernández-Mayoralas, A.; Jiménez-Barbero, J.; Martin-Lomas, M. J. Org. Chem. **1995**, 60, 1502. (c) Ichikawa, Y.; Lin, Y.-C.; Dumas, D. P.; Shen, G. J.; Garcia-Junceda, E.; Williams, M. A.; Bayer, R.; Ketcham, C.; Walker, L. E.; Paulson, J. C.; Wong, C.-H. J. Am. Chem. Soc. **1992**, 114, 9283. See also: (d) Bajorath, J.; Stenkamp, R.; Aruffo, A. Bioconjugate Chem. **1995**, 6, 3 and references therein.

(11) See also: (a) Scheffer, K.; Ernst, B.; Katopodis, A.; Magnani, J. L.; Wang, W. T.; Weisemann, R.; Peters, T. *Angew. Chem., Int. Ed. Engl.* **1995**, *34*, 1841; (b) Cooke, R. M.; Hale, R. S.; Lister, S. G.; Shal, G.; Weir, M. P. *Biochemistry* **1994**, *33*, 10591. (c) Ball, G. E.; O'Neill, R. A.; Schultz, J. E.; Lowe, J. B.; Weston, B. W.; Nagy, J. O.; Brown, E. G.; Hobbs, C. J.; Bednarski, M. D. J. Am. Chem. Soc. **1992**, *114*, 5449. (d) Lin, Y.-C.; Hummel, C. W.; Huang, D.-H.; Ichikawa, Y.; Nicolaou, K. C.; Wong, C.-H. J. Am. Chem. Soc. **1992**, *114*, 5452.

Nicolaou, K. C.; Wong, C.-H. J. Am. Chem. Soc. **1992**, 114, 5452. (12) (a) Somers, W. S.; Tang, J.; Shaw, G. D.; Camphausen, R. T. Cell **2000**, 103, 467. (b) Graves, B. J.; Crowther, R. L.; Chandran, C.; Rumberger, J. M.; Lis, S.; Huang, K.-S.; Presky, D. H.; Familletti, P. C.; Wolitsky, B. A.; Burns, D. K. Nature **1994**, 367, 532.

(13) Toepfer, A.; Kretschmar, G.; Bartnik, E. Tetrahedron Lett. 1995, 36, 1161.

(14) (a) Kolb, H. C.; Ernst, B. *Chem. Eur. J.* **1997**, 1571. (b) Janke, W.; Kolb, H. C.; Blommers M. J. J.; Magnani, J. L.; Ernst, B. *Angew. Chem., Int. Ed. Engl.* **1997**, *36*, 2603. (c) Kolb, H. C.; Ernst, B. *Pure Appl. Chem.* **1997**, *69*, 1879.

⁽¹⁵⁾ Janke, W.; Kolb, H. C.; Blommers, M. J. J.; Magnani, J. L.; Ernst, B. Angew. Chem., Int. Ed. Engl. **1997**, *36*, 2603.

(16) (a) Thoma, G.; Magnani, J. L.; Patton, J. T.; Ernst, B.; Janke,
W. Angew. Chem., Int. Ed. Engl. 2001, 40, 1941. (b) Thoma, G.; Bänteli,
R.; Jahnke, W.; Magnani, J. L.; Patton, J. T. Angew. Chem., Int. Ed.
2001, 40, 3644. (c) Thoma, G.; Duthaler, R. O.; Magnani, J. L.; Patton,
J. T. J. Am. Chem. Soc. 2001, 123, 10113.

(17) (a) Foster, A. B.; Stacey, M.; Vardheim, S. V. Acta Chem. Scand.
 1958, 12, 1819. (b) Czernecki, S.; Vijayakumaran, K.; Ville, G. J. Org. Chem. **1986**, 51, 5472. (c) Giuliano, R. M.; Jordan, A. D., Jr.; Gauthiere, A. D.; Hoogsteen, K. J. Org. Chem. **1993**, 58, 4979.



Figure 1. Some E-selectin carbohydrate-based antagonists (refs 16 and 18).

preorganization of the bioactive conformation of $\mathbf{2}$, as well as multivalency was recently confirmed.¹⁶

The synthesis and biological evaluation of **2** and related analogues was reported by Thoma and co-workers¹⁸ at Novartis Pharma in Basel, jointly with Patton and Magnani of Glycotech. Their synthesis strategy as illustrated in Figure 2 relies on glycosylations with D-gal¹⁹ and α -L-fucosyl²⁰ thioglycosides. The crucial α -glycolate ether linkage was achieved by treatment of the 3,4-

^{(9) (}a) Tyrrell, D.; James, P.: Rao, N.; Foxall, Cl; Abbas, S.; Dasgupta, F.; Nashed, M.; Hasegawa, A.; Kiso, M.; Asa, D.; Kidd, J.; Brandley, B. K. *Proc. Natl. Acad. Sci. U.S.A.* **1991**, *88*, 10372. (b) Ramphal, J. T.; Zheng, Z.-L.; Perez, C.; Walker, L. E.; DeFrees, S. A.; Gaeta, F. C. A. *J. Med. Chem.* **1994**, *37*, 3459. (c) Stahl, W.; Sprengard, U.; Kretschmar, G.; Kunz, H. *Angew. Chem., Int. Ed. Engl.* **1994**, *33*, 2096.

⁽¹⁸⁾ Thoma, G.; Kinzy, W.; Bruns, C.; Patton, J. T.; Magnani, J. L.; Bänteli, R. *J. Med. Chem.* **1999**, *42*, 4909.

⁽¹⁹⁾ Contour, M. O.; Defaye, J.; Little, M.; Wong, E. *Carbohydr. Res.* **1989**, *193*, 283.

⁽²⁰⁾ Yamazaki, F.; Kitajima, T.; Nukada, T.; Ito, T.; Ogawa, T. Carbohydr. Res. 1990, 201, 15.



Figure 2. Novartis Pharma disconnection of antagonist 2 (ref 18).

stannylidene acetal derived from the D-gal unit with the triflate ester prepared from methyl 3-cyclohexyl-2*R*-hydroxy propionate.²¹ This regioselective activation of vicinal diols via organotin derivatives in S_N^2 -type etherifications is well-known.²² Further elaboration afforded the intended target **2**, which proved to be approximately 30 times more active than sialyl Lewis^x in a static cell-free E-selectin binding assay.

Our interest in the synthesis of **2** was motivated by a number of considerations and challenges. Foremost among these was the need to explore methodology that did not utilize thioglycosides²³ or other relatively labile activating groups²⁴ as glycosyl donors. Another factor was to avoid organotin chemistry late in the synthesis, to render a potential scale-up of the process practical.²⁵ We considered these issues as a good opportunity to apply recently developed glycosylation methods that rely on the remote activation concept²⁶ utilizing shelf-stable and often crystalline glycosyl donors. The exploration of

alternative methods for ether formation, not relying on tin-mediated activation of hydroxyl groups was another challenge.

Figure 3 shows critical disconnections starting with **2**, where glycoside formation relies on a stereocontrolled α -L-fucosylation and β -D-galactosylation reactions with a common 2-thiopyridyl carbonate (TOPCAT) leaving group.²⁷ Etherification would involve a Rh^{II}-catalyzed carbene insertion reaction,²⁸ followed by a Horner–Wadsworth–Emmons extension and catalytic hydrogenation.

The readily available preferentially protected trimethylsilylethyl β -D-galactopyranoside **3**²⁹ was treated with methyl diazo(dimethoxyphosphoryl)acetate³⁰ under Rh^{II} catalysis³¹ to afford the 3-O- α -phosphonoacetate 5 as a mixture of epimers in 72% yield (Scheme 1). Olefination with cyclohexane carboxyaldehyde in the presence of DBU and lithium chloride³² afforded **6** as a major isomer in a 9:1 mixture in 97% yield as evidenced by NOE studies. Hydrogenation with 20% palladium hydroxide on carbon³³ (Pearlman's catalyst, Degussa), followed by treatment with benzaldehyde dimethylacetal in the presence of fluoroboric $acid^{34}$ led to a single isomer 7. Benzoylation afforded a crystalline product 8, suitable for single-crystal X-ray analysis as shown in the ORTEP drawing in Scheme 1. Thus, the correct stereochemistry at the α -cyclohexyl glycolate carbon atom was secured.

Scheme 2 shows the conversion of **8** into the required activated glycosyl donor **11**. In preliminary studies, several conditions normally utilized to cleave silyl glycosides such as TBAF/AcOH, TBAF/THF, or CsF/DMF, with or without 18-crown-6, resulted in recovery of starting material, even when heated (i.e., CsF/DMF, MeCN, or THF). Cleavage was successfully achieved with TFA in CH₂Cl₂ at 0 °C within 20 min, which also led to the hydrolysis of the benzylidene acetal. Subsequent steps involved acetal formation to **10** and esterification²⁷ with di(*S*-2-pyridyl) thiocarbonate³⁵ at the anomeric center to give the glycosyl donor **11** as a pale yellow solid in excellent yield.

The acceptor disaccharide unit was prepared form the readily available 1,5-anhydro-D-*xylo*-hexitol derivative 13^{17} (Scheme 3). Benzylation afforded 14, which was subjected to reductive cleavage³⁶ of the 4,6-*O*-benzylidene acetal to afford the 6-*O*-benzyl derivative 15. Glycosylation with L-fucosyl 2-thiopyridyl carbonate donor 16 proceeded in nearly quantitative yield to give the desired α -L-fucosylated disaccharide 17. Subsequent debenzoylation gave the desired acceptor 18.

(34) Albert, R.; Dax, K.; Pleschko, R.; Stütz, A. E. Carbohydr. Res. 1985, 137, 282.

(35) Corey, E. J.; Clark, D. A. Tetrahedron Lett. 1979, 2875.

(36) Garegg, P.; Hultberg, H.; Wallin, S. *Carbohydr. Res.* **1982**, *108*, 97.

⁽²¹⁾ Degerbeck, F.; Frausson, B.; Grehm, L.; Ragnarsson, J. Chem. Soc., Perkin Trans. 1 1993, 11.

^{(22) (}a) Nashed, M. A.; Anderson, L. *Tetrahedron Lett.* **1976**, *39*, 3503. (b) Nagashima, N.; Ohno, M. *Chem. Lett.* **1987**, 141. For a review, see: (c) David, S.; Hanessian, S. *Tetrahedron* **1985**, *41*, 643.

⁽²³⁾ For reviews, see: (a) Norberg, T. In Modern Methods in Carbohydrate Synthesis; Khan, S. H., O'Neill, R. A., Eds.; Harwood Academic: Chichester, 1996; Chapter 4, p 82. (b) Garegg, P. Adv. Carbohydr. Chem. Biochem. **1997**, 52, 179.

⁽²⁴⁾ For reviews, see: *Preparative Carbohydrate Chemistry*, Hanessian, S., Ed.; Marcel Dekker: New York, 1997; Chapter 12–14.

⁽²⁵⁾ For a discussion on large-scale synthesis of sialyl Lewis^x, see: Kretschmar, G.; Stahl, W. *Tetrahedron* **1998**, *54*, 6341.

⁽²⁶⁾ For a review, see: (a) Hanessian, S.; Lou, B. *Chem. Rev.* **2000**, *100*, 4443. Hanessian, S. In *Preparative Carbohydrate Chemistry*; Hanessian, S., Ed.; Marcel Dekker: New York, 1997; Chapter 16, p 381.

⁽²⁷⁾ See, for example: (a) Hanessian, S.; Huynh, H. K.; Reddy, G. V.; Duthaler, R. O.; Katopodis, A.; Streiff, M. B.; Kinzy, W.; Ohrlein, R. *Tetrahedron* 2001, *57*, 3281. (b) Lou, B.; Huynh, H. K.; Hanessian, S. In *Preparative Carbohydrate Chemistry*; Hanessian, S., Ed.; Marcel Dekker: New York, 1997; Chapter 19, p 431.
(28) For reviews, see: (a) Miller, D. J.; Moody, C. J. *Tetrahedron*

⁽²⁸⁾ For reviews, see: (a) Miller, D. J.; Moody, C. J. Tetrahedron 1995, 51, 10811. (b) Kirmse, W. In Advances in Carbene Chemistry; Brinker, V., Ed.; JAI Press: Greenwich, CT, 1994; Vol. 1, p 1. (c) Adams, J.; Spero, D. M. Tetrahedron 1991, 47, 1769.

⁽²⁹⁾ Nilsson, U.; Ray, A. K.; Magnusson, G. Carbohydr. Res. 1994, 252, 117.

⁽³⁰⁾ Kane, A. B.; McKenna, C. E. Synthesis 1991, 409.

⁽³¹⁾ See, for example: Paquet, E.; Sinay, P. Tetrahedron Lett. 1984, 25, 3071.

 ⁽³²⁾ Blanchette, M. A.; Choy, N.; Davis, J. T.; Essenfeld, A.;
 Masamune S.; Roush, W. R.; Sakai, T. *Tetrahedron Lett.* **1984**, *29*, 2183.
 (33) Pearlman, W. M. *Tetrahedron Lett.* **1967**, 1663.



Figure 3. Disconnective analysis.

The β -D-galactosylation reaction required careful experimentation no doubt due to the hindered nature of the hydroxyl group in 18. Under previously successful conditions using silver triflate³⁷ as promoter in the presence of 4 Å molecular sieves, and tetramethylurea^{37a,b} as an acid scavenger, only low yields of the expected trisaccharide 19 were obtained. Eventually, it was found that pretreatment of the acceptor 18 with sodium hydride, followed by addition of silver triflate, and finally adding the donor 11 led to the target trisaccharide 19 in \sim 50% yield (Scheme 4). Cleavage of the methyl ester afforded the corresponding carboxylic acid 20 in quantitative yield. Debenzoylation required refluxing in methanol in the presence of sodium methoxide affording the penultimate precursor 21 in excellent yield. Removal of the benzyl ethers utilizing Pearlman's catalyst³³ under 60 psi of hydrogen in a mixture of aqueous dioxane containing acetic acid, followed by conversion to the sodium salt, afforded the target inhibitor 2 in quantitative yield.

We have described a new and practical synthesis of the potent carbohydrate-based E-selectin inhibitor **2**, utilizing a strategy that avoids thioglycosides as glycosyl donors and organotin activation late in the synthesis. The combined yields of α -L-fucosylation and β -D-galactosylation reactions utilizing thioethyl anomeric activation in the Novartis synthesis was 73%.¹⁸ Coincidentally the yields of two glycosylations utilizing the 2-thiopyridyl carbonate group in the synthesis described herein was also 73%. The critical etherification reaction in our synthesis relied on the generation of a vinyl ether and a highly stereoselective catalytic hydrogenation proceeding in a combined yield of ~90%. Overall, the synthesis of **2** is convergent, high yielding in many steps, with the added advantage of a crystalline advanced intermediate, thus adding an element of practicality.

Experimental Section

Solvents were distilled under a positive pressure of dry nitrogen before use and dried by standard methods; THF and ether, from Na/benzophenone; and CH₂Cl₂, from CaH₂. All commercially available reagents were used without further purification. All reactions were performed under nitrogen atmosphere. NMR (¹H, ¹³C) spectra were recorded on AMX-300 and ARX-400 spectrometers. The term [(-)] in ¹³C data refers to the sign of the corresponding peak in the DEPT 135 NMR experiment. Low- and high-resolution mass spectra were recorded on Finningan MAT 900, VG Micromass, Ael-MS902, or Kratos MS-50 spectrometers using fast atom bombardment (FAB) or electrospray techniques. Optical rotations were recorded on a Perkin-Elmer 241 polarimeter in a 1 dm cell at ambient temperature. Analytical thin-layer chromatography was performed on Merck 60F₂₅₄ precoated silica gel plates. Visualization was done by ultraviolet light and/or by staining with ceric ammonium molybdate. Flash column chromatography³⁸ was performed using (40–60 μ M) silica gel at increased pressure.

Methyl Diazo(dimethoxyphosphoryl)acetate (4).³⁰ To a suspension of *t*-BuOK (dried overnight under vacuum over

^{(37) (}a) Hanessian, S.; Banoub, J. *Carbohydr. Res.* 1977, *53*, C13.
(b) *Methods. Carbohydr. Chem.* 1980, *8*, 243. See also: (c) Schuerch, C.; Kronzer, F. J. *Carbohydr. Res.* 1973, *27*, 379. For a recent application to glycopeptides, see: Polt, R.; Szabó, L.; Treiberg, J.; Li, Y.; Hruby, V. J. *J. Am. Chem. Soc.* 1992, *114*, 10249.



 $P_2O_5)$ (5.17 g, 46.2 mmol) in toluene (200 mL) cooled at 0 °C was added dropwise a solution of methyl (dimethoxyphosphoryl)acetate (7.00 g, 38.5 mmol) in toluene (20 mL) while maintaining the temperature below 5 °C. The viscous mixture was stirred for 1 h, and a solution of naphthalenesulfonyl azide³⁰ (9.05 g, 38.8 mmol) in toluene (20 mL) was added

80%



dropwise at 5 °C. After further stirring for 2 h, the mixture was filtered and concentrated, and the residue was purified by flash chromatography on neutral alumina, eluting with benzene/hexanes 1:1 to give **4** (3.9 g, 54%): bp 110 °C/10 mmHg (lit.³⁰ bp 65–69 °C/0.002 Torr); ¹H NMR (CDCl₃, 400 MHz) δ 3.84 (d, 3H, CO₂CH₃), 3.81 (d, 6H, P(OCH₃)₂, ³J_{H-P} = 3.5 Hz); ¹³C NMR (CDCl₃, 100 MHz) δ 163.5 (d, CO, ²J_{C-P} = 12.5 Hz), 128.4 (d, CN₂, ¹J_{C-P} = 81 Hz), 53.7 (d, P(OCH₃)₂, ³J_{Z-P} = $^{2}J_{C-P} = 6$ Hz), 52.5 (s, OCH₃); ³¹P NMR (CDCl₃, 162 MHz) δ 13.8; IR (film, cm⁻¹) 2134, 1711, 1288, 1029; HR-FABMS calcd for C₅H₉O₅N₂PNa *m*/*z* 231.01468, found 231.01465.

2-(Trimethylsilyl)ethyl *O*-2,4,6-Tri-*O*-benzyl-3-*O*-[(*E*)-1-(methoxycarbonyl)-2-cyclohexylethylen-1-oxy]- β -D-galactopyranoside (6). A mixture of 2-(trimethylsilyl)ethyl 2,4,6-tri-*O*-benzyl- β -D-galactopyranoside²⁹ 3 (4.67 g, 8.48 mmol), methyl diazo(dimethoxyphosphoryl)acetate 4 (3.53 g, 2 equiv), and rhodium(II) acetate dimer (150 mg, 2 mol %/4) in benzene (280 mL) was refluxed for 5 h. After filtration on Celite and concentration, the residue was purified by flash chromatog-raphy on silica gel, eluting with ethyl acetate/hexanes 35:65, to give a 1:1 diastereomeric mixture of phosphonates 5 and 5' (4.43 g, 71.5%): HR-FABMS calcd for C₃₇H₅₁O₁₁PSiNa *m*/*z* 753.28360, found 753.28296.

A solution of the above mixture (2.80 g, 3.82 mmol), lithium chloride (dried under vacuum over P2O5 before use) (195 mg, 1.2 equiv), and DBU (distilled under reduced pressure over CaH_2 before use) (630 μ L, 1.1 equiv) dissolved in acetonitrile (38 mL) was stirred at rt for 30 min, and then cyclohexane carboxaldehyde (510 μ L, 1.1 equiv) was added dropwise at 0 °C and the mixture stirred at this temperature for 15 min (formation of a vellow precipitate). The resulting mixture was concentrated, redissolved in dichloromethane (50 mL), and washed with saturated aqueous ammonium chloride (40 mL) and 1 M aqueous hydrogen chloride (20 mL). The organic phase was dried (sodium sulfate) and concentrated. The residue was purified by flash chromatography on silica gel, eluting with ethyl acetate/hexanes 1:9, to give a 9:1 inseparable mixture of isomers (2.65 g, 97%). Data for the major isomer 6: $[\alpha]_D$ -13.8 (c 1.0, CHCl₃); ¹H NMR (CDCl₃, 400 MHz, assigned by COSY45) δ 7.41–7.28 (m, 15H, ArH), 5.51 (d, 1H, H_{vinylic}, J= 9.9 Hz), 4.98 (d, 1H, PhCHH-, J = 11.7 Hz), 4.93 (d, 1H, PhC*H*H-, *J* = 10.8 Hz), 4.73 (d, 1H, PhCH*H*-, *J* = 10.8 Hz), 4.64 (d, 1H, PhCHH-, J = 11.7 Hz), 4.46 and 4.41 (AB, 2H, PhC H_2 -, J = 11.7 Hz), 4.41 (d, 1H, H-1, $J_{1,2} = 7.6$ Hz), 4.08-4.00 (m, 2H, H-4 and -OCHHCH2TMS), 3.97 (dd, 1H, H-3, $J_{3,2} = 9.7$ Hz, $J_{3,4} = 2.9$ Hz), 3.86 (dd, 1H, H-2), 3.75 (s, 3H, OCH₃), 3.63-3.55 (m, 4H, H-5, H-6, H-6A and -OCHHCH₂-TMS), 2.84 (m, 1H, -CH-), 1.71-1.68 (m, 6H, -CH₂-), 1.36- $0.98 (2m, 6H, -CH_2 - and -CH_2TMS), 0.03 (s, 9H, -Si(CH_3)_3);$ $^{13}\mathrm{C}$ NMR (CDCl_3, 100 MHz) δ 164 (CO), 143.8, 138.9 (2C) and 138.1 (ArC), 128.5, 128.3, 128.2, 128.1, 128.0, 127.9, 127.8,

Scheme 4



127.6 and 127.4 (ArCH), 103.5 (C-1), 82.7, 80.0, 73.8, 73.4 (C-2 to C-5), 75.0, 74.6, 73.5 [(-), PhCH₂-], 69.0 and 67.6 [(-), C-6 and $-OCH_2CH_2TMS$], 51.7 ($-OCH_3$), 36.0 (-CH-), 33.7, 33.5, 26.0, 25.9 (2C) [(-), $-CH_2$ -], 18.6 [(-), $-CH_2TMS$], -1.3 [$-Si-(CH_3)_3$]; IR (film, cm⁻¹) 2926, 2852, 1724, 1249, 1216, 1102, 1076; HR-FABMS calcd for C₄₂H₅₆O₈SiNa *m*/*z* 739.36422, found 739.36364.

2-(Trimethylsilyl)ethyl O-4,6-O-Benzylidene-3-O-[(S)-1-(methoxycarbonyl)-2-cyclohexylethyloxy]-β-D-galactopyranoside (7). To a solution of 6 and its inseparable isomer (1.07 g, 1.49 mmol) dissolved in methanol (21 mL) was added palladium hydroxide on carbon (Degussa, 107 mg). The resulting mixture was hydrogenated under 60 psi at rt for 2 h. After filtration on Celite and concentration, the residue was dissolved in DMF (15 mL) and cooled with an ice bath, and then benzaldehyde dimethyl acetal (268 $\mu L,$ 1.2 equiv) and tetrafluoroboric acid (54% in ether, 246 μ L, 1.2 equiv) were added.³⁴ The mixture was stirred under nitrogen atmosphere overnight at rt, and then triethylamine (270 μ L, 1.3 equiv) was added and the solvent evaporated. The residue was purified by flash chromatography on silica gel, eluting with ethyl acetate/hexanes 2:8 to give 7 (640 mg, 80%) as clear syrup: [α]_D –18.1 (c 0.95, CHCl₃); ¹H NMR (CDCl₃, 400 MHz, assigned by COSY45) δ 7.51 (d, 2H, ArH, J = 7.1 Hz), 7.34–7.30 (m, 3H, ArH), 5.53 (s, 1H, PhCH(O)(O)-), 4.50 (dd, 1H, -OCHCO₂-Me, J = 9.4, 3.6 Hz), 4.34 (fd, 1H, H-4, $J_{4,3} = 3.5$ Hz), 4.30 (d, 1H, H-6, $J_{6.6A} = 12.4$ Hz), 4.26 (d, 1H, H-1, $J_{1,2} = 7.8$ Hz), 4.07 (d, 1H, H-6A), 4.05 (m, 1H, -OCHHCH2TMS), 3.95 (dd, 1H, H-2, $J_{2,3} = 9.6$ Hz), 3.62 (s, 3H, OCH₃), 3.55 (m, 1H, -OCHHCH2TMS), 3.48 (dd, 1H, H-3), 3.39 (brs, 1H, H-5), 2.50 (brs, 1H, -OH), 1.90-1.50 (3m, 9H, -CH₂- and -CH-), 1.30-0.80 (m, 6H, -CH₂- and -CH₂TMS), 0.00 (s, 9H, -Si-(CH₃)₃); ¹³C NMR (CDCl₃, 100 MHz) δ 174.7 (CO), 138,0 (ArC), 128.9, 128.0 (2C) and 126.6 (2C) (ArCH), 102.7 and 101.0 (PhCH(O)(O)- and C-1), 79.4, 77.0, 74.9, 70.8 and 66.6 (C-2 to C-5 and -OCHCO2Me), 69.3 and 67.4 [(-), C-6 and -OCH2-CH₂TMS], 51.9 (-OCH₃), 40.8, 33.9, 32.6, 26.6, 26.4 and 26.3 [(-), -CH₂-], 33.9 (-CH-), 18.3 [(-), -CH₂TMS], -1.2 [-Si-(CH₃)₃]; IR (film, cm⁻¹) 2923, 1741, 1249, 1113, 1051; HR-FABMS calcd for C₂₈H₄₄O₈SiNa m/z 559.27032, found 559.27057.

Lactone Derivative of 2-(Trimethylsilyl)ethyl *O*-4,6-*O*-Benzylidene-3-*O*-[(*S*)-1-(methoxycarbonyl)-2-cyclohexylethyloxy]- β -D-galactopyranoside (9). To a cooled solution of 7 (112 mg, 208.1 μ mol) in THF (21 mL) was added at 0 °C lithium hydroxide (17.5 mg, 2 equiv) dissolved in water (2 mL). The mixture was stirred overnight at 7 °C (cold chamber) and neutralized with Amberlite IR-120 (H⁺). After filtration and concentration, the residue was dissolved in a mixture of acetic anhydride (1.2 mL) and acetonitrile (4.6 mL)³⁹ and stirred at

rt for 12 h. After concentration, the residue was purified by flash chromatography on silica gel, eluting with ethyl acetate/ hexanes 2:8, to give 9 (104 mg, 99%) as a syrup: $[\alpha]_D -37$ (c 1.0, CHCl₃); ¹H NMR (CDCl₃, 400 MHz, assigned by COSY45) δ 7.50 (d, 2H, ArH, J= 7.1 Hz), 7.35 (m, 3H, ArH), 5.59 (s, 1H, PhCH(O)(O)-), 4.66 (dd, 1H, H-2, $J_{2,3} = 10.1$ Hz, $J_{2,1} =$ 7.7 Hz), 4.58 (m, 1H, -OCHCO2Me), 4.57 (d, 1H, H-1), 4.37 (fdd, 1H, H-6, J_{6,6A} = 12.5 Hz, J_{6,5} = 1.3 Hz), 4.33 (fd, 1H, H-4, $J_{4,3} = 3.1$ Hz), 4.11 (fdd, 1H, H-6A, $J_{6A,5} = 1.3$ Hz), 4.10 (m, 1H, -OCHHCH2TMS), 3.78 (dd, 1H, H-3), 3.63 (m, 1H, -OCHHCH2TMS), 3.53 (brs, 1H, H-5), 1.88-1.50 (m, 9H, -CH₂- and -CH-), 1.30-0.80 (m, 6H, -CH₂- and -CH₂-TMS), 0.03 (s, 9H, $-Si(CH_3)_3$); ¹³C NMR (CDCl₃, 100 MHz) δ 170.2 (CO), 137,5 (ArC), 129.3, 128.4 (2C) and 126.5 (2C) (ArCH), 101.2 and 100.0 (PhCH(O)(O)- and C-1), 74.5, 74.1, 72.3, 71.4 and 66.8 (C-2 to C-5 and -OCHCO2Me), 69.3 and 67.7 [(-), C-6 and -OCH2CH2TMS], 38.6, 34.1, 32.1, 26.6, 26.4 and 26.1 [(-), -CH₂-], 33.7 (-CH-), 18.2 [(-), -CH₂TMS], -1.2 [-Si(CH₃)₃]; IR (film, cm⁻¹) 2925, 2853, 1756, 1251, 1049; HR-FABMS calcd for C₂₇H₄₀O₇SiNa m/z 527.24410, found 527.24440.

2-(Trimethylsilyl)ethyl O-2-O-Benzoyl-4,6-O-benzylidene-3-O-[(S)-1-(methoxycarbonyl)-2-cyclohexylethyloxy]- β -**D**-galactopyranoside (8). To a cold solution of 7 (300 mg, 559 µmol) in dichloromethane (5.6 mL) and pyridine (1.1 mL) were added benzoyl chloride (325 μ L, 5 equiv) and a catalytic amount of DMAP at 0 °C. The mixture was stirred at rt overnight. The excess of benzoyl chloride was quenched with MeOH and the mixture concentrated under vacuum. The crude product was redissolved in dichloromethane (25 mL), washed with aqueous saturated ammonium chloride (20 mL) and water (20 mL), and dried (sodium sulfate). After concentration, the residue was purified by flash chromatography on silica gel, eluting with ethyl acetate/hexanes 3:7, to give 8 (359 mg, 100%) as a white solid. White needle-shaped crystals were obtained by recrystallization from a mixture of hexanes/ethyl acetate: mp 160 °C; $[\alpha]_D$ +10 (*c* 0.2, CHCl₃); ¹H NMR (CDCl₃, 400 MHz, assigned by COSY45) δ 8.08 (d, 2H, ArH, J = 7.1Hz), 7.60–7.30 (3m, 8H, ArH), 5.61 (dd, 1H, H-2, $J_{2,3} = 10.0$ Hz, $J_{2,1} = 8.0$ Hz), 5.58 (s, 1H, PhCH(O)(O)-), 4.60 (d, 1H, H-1), 4.46 (fd, 1H, H-4, J_{4,3} = 3.5 Hz), 4.35 (fdd, 1H, H-6, J_{6,6A} = 12.3 Hz, $J_{6,5}$ = 1.5 Hz), 4.13 (m, 1H, $-OCHCO_2Me$), 4.11 (fdd, 1H, H-6A, $J_{6A,5} = 1.5$ Hz), 4.00 (m, 1H, $-OCHHCH_2TMS$), 3.76 (dd, 1H, H-3), 3.58 (s, 3H, OCH₃), 3.54 (m, 1H, -OCHHCH₂-TMS), 3.48 (brs, 1H, H-5), 1.57 (m, 1H, -CHHCHCO₂Me-), 1.50-1.20 (m, 7H, -CH₂- and -CH*H*CHCO₂Me-), 0.95-0.75 (m, 5H, -CH₂-, -CH- and -CH₂TMS), 0.71-0.55 (m, 2H, -CH₂-), 0.08 (s, 9H, -Si(CH₃)₃); ¹³C NMR (CDCl₃, 100 MHz) δ 174.3 (CO₂Me), 165.0 (PhCO), 138.0 and 130.4 (ArC), 133.1, 130.0 (2C), 129.0, 128.5 (2C), 128.2 (2C) and 126.7 (2C) (ArCH), 101.2 and 100.9 (PhCH(O)(O)- and C-1), 79.0, 78.3, 75.2, 71.9 and 66.9 (C-2 to C-5 and -OCHCO2Me), 69.2 and 66.8 [(-),

^{(39) (}a) Zou, W.; Jennings, H. J. J. Carbohydr. Chem. 1996, 15, 257.
(b) Severn, W. B.; Richards, C. J. Am. Chem. Soc. 1993, 115, 1114.

C-6 and $-OCH_2CH_2TMS$], 52.0 ($-OCH_3$), 41.0, 33.8, 32.6, 26.4, 25.9 and 25.7 [(-), $-CH_2-$], 33.3 (-CH-), 18.0 [(-), $-CH_2-$ TMS], -1.3 [$-Si(CH_3)_3$]; IR (film, cm⁻¹) 2924, 1731, 1267, 1096; HR-FABMS calcd for $C_{35}H_{48}O_9SiNa$ *m*/*z* 663.29653, found 663.29590.

2-O-Benzoyl-4,6-O-benzylidene-3-O-[(S)-1-(methoxycarbonyl)-2-cyclohexylethyloxy]-a-D-galactopyranose (10). A solution of **8** (200 mg, 312 μ mol) in dichloromethane (1.5 mL) and TFA (3.0 mL) was stirred at 0 °C for 30 min, concentrated, and codistilled twice with toluene. The crude product was dissolved in DMF (3.1 mL) cooled in an ice bath, and then benzaldehyde dimethyl acetal (56 μ L, 1.2 equiv) and tetrafluoroboric acid (54% in ether, 47 μ L, 1.1 equiv) were added. The mixture was stirred under nitrogen atmosphere for 6 h at rt, triethylamine (52 μ L, 1.2 equiv) was added, and the solvent was evaporated. The residue was purified by flash chromatography on silica gel, eluting with ethyl acetate/ hexanes 1:1, to give a 4:1 α/β inseparable mixture of isomers (115 mg, 68%) as a clear syrup. Data for the major α isomer 10: ¹H NMR (CDCl₃, 400 MHz, assigned by COSY45) δ 8.08 (d, 2H, ArH, J = 7.3 Hz), 7.59-7.55 (m, 3H, ArH), 7.48-7.32 (2m, 5H, ArH), 5.67 (d, 1H, H-1, $J_{1,2} = 3.5$ Hz), 5.58 (s, 1H, PhCH(O)(O)-), 5.54 (dd, 1H, H-2, $J_{2,3} = 10.4$ Hz), 4.53 (fd, 1H, H-4, $J_{4,3} = 3.2$ Hz), 4.31 (dd, 1H, $-OCHCO_2Me$, J = 9.1, 4.0 Hz), 4.23 (d, 1H, H-6, $J_{6,6A} = 12.5$ Hz), 4.12 (dd, 1H, H-3), 4.06 (d, 1H, H-6A), 3.94 (brs, 1H, H-5), 3.64 (s, 3H, OCH₃), 1.62-1.27 (m, 8H, -CH₂- and -CHHCHCO₂Me-), 1.00-0.60 (m, 5H, $-CH_2-$, -CH-); ¹³C NMR (CDCl₃, 100 MHz) δ 174.5 (CO2Me), 166.0 (PhCO), 137.9, 133.4, 129.9, 129.6 (2C), 129.0, 128.6 (2C), 128.2 (2C) and 126.5 (2C) (ArCH and ArC), 100.9 (PhCH(O)(O)-), 91.2 (C-1), 78.4, 75.7, 74.2, 71.8, 69.4 and 62.9 (C-2 to C-6 and -OCHCO2Me), 52.1 (-OCH3), 41.0, 33.8, 33.7, 32.8, 26.4, 26.1 and 25.8 (-CH2- and -CH-); HR-FABMS calcd for C₃₀H₃₆O₉Na m/z 563.22570, found 563.22667.

2-O-Benzoyl-4,6-O-benzylidene-3-O-[(S)-1-(methoxycarbonyl)-2-cyclohexylethyloxy]- β -D-galactopyranosyl **2-Thiopyridyl Carbonate (11).** A mixture of **10** and its β isomer (99.3 mg, 183.7 µmol), di(S-2-pyridyl) thiocarbonate (137 mg, 3 equiv), and triethylamine (77 µL, 3 equiv) in dichloromethane (1.84 mL) was stirred at rt for 24 h. Concentration and purification by flash chromatography on silica gel, eluting with ethyl acetate/hexanes 1:1, gave 11 (100 mg, 80%) as a yellow powder: $[\alpha]_D$ +17.3 (*c* 1.1, CHCl₃); ¹H NMR (CDCl₃, 400 MHz, assigned by COSY45) δ 8.52 (m, 1H, PyH), 8.07 (d, 2H, ArH, J = 7.1 Hz), 7.70-7.30 (4m, 10H, ArH), 7.25 (m, 1H, PyH), 5.87 (d, 1H, H-1, $J_{1,2} = 8.3$ Hz), 5.80 (dd, 1H, H-2, $J_{2,3}$ = 9.8 Hz), 5.61 (s, 1H, PhCH(O)(O)-), 4.56 (fd, 1H, H-4, J_{4,3} = 3.2 Hz), 4.37 (dfd, 1H, H-6, $J_{6,6A}$ = 12.5 Hz, $J_{6,5}$ = 1.3 Hz), 4.15 (m, 1H, $-OCHCO_2Me$), 4.12 (dfd, 1H, H-6A, $J_{6A,5}$ = 1.3 Hz), 3.81 (dd, 1H, H-3), 3.65 (brs, 1H, H-5), 3.63 (s, 3H, OCH₃), 1.59 (m, 1H, -CHHCHCO2Me-), 1.50-1.20 (m, 7H, -CH2and $-CHHCHCO_2Me-$), 1.00-0.58 (2m, 5H, $-CH_2-$, -CH-); ¹³C NMR (CDCl₃, 100 MHz) δ 174.0 (CO_2Me), 168.6 and (-OC(O)S-), 164.7 (PhCO), 151.2 (PyC), 150.6 (PyCH), 137.9, 137.7, 133.8, 130.3, 130.0, 129.7, 129.3, 128.9, 128.5, 126.8 and 124.0 (PhCH PyCH and PhC), 101.4 and 95.2 (PhCH(O)(O)and C-1), 79.0, 78.9, 74.9, 70.5 and 68.2 (C-2 to C-5 and -OCHCO₂Me), 68.9 [(-), C-6], 52.3 (-OCH₃), 41.1, 34.0, 32.9, 26.6, 26.2 and 25.9 [(-), -CH₂-], 33.6 (-CH-); IR (film, cm⁻¹) 2924, 1736, 1264, 1112, 1062; HR-FABMS calcd for C35H39O8-NSNa m/z 656.22940, found 656.22971.

3-*O*-Benzoyl-4,6-*O*-benzylidene-1,2-dideoxy-D-glucopyranose (14). To a cold (5 °C) solution of 4,6-*O*-benzylidene-1,2-dideoxy-D-glucopyranoside¹⁷ **13** (1.0 g, 4.24 mmol) in dichloromethane (47 mL) were added triethylamine (1.77 mL, 3 equiv), a catalytic amount of DMAP, and benzoyl chloride (1.0 mL, 2 equiv). The mixture was stirred at rt for 2 h, poured into a separatory funnel, washed with water and brine, dried (sodium sulfate), and filtered. After concentration, the residue was purified by flash chromatography on silica gel, eluting with ethyl acetate/hexanes 1:9, to give **14** (1.40 g, 97%) as a clear syrup: $[\alpha]_D - 135.8$ (*c* 0.5, CHCl₃); ¹H NMR (CDCl₃, 400 MHz, assigned by COSY45) δ 8.08 (d, 2H, ArH, *J* = 7.0 Hz), 7.57-7.33 (3m, 8H, ArH), 5.63 (s, 1H, PhC*H*(O)(O)-), 5.36 (ddd, 1H, H-3, *J*_{3,2ax} = 10.9 Hz, *J*_{3,4} = 9.5 Hz, *J*_{3,2eq} = 5.3 Hz), 4.36 (dd, 1H, H-6, $J_{6,6A} = 10.4$ Hz, $J_{6,5} = 4.9$ Hz), 4.06 (ddfd, 1H, H-1eq, $J_{1eq,1ax} = 12.0$ Hz, $J_{1eq,2ax} = 5.2$ Hz, $J_{1eq,2eq} = 1.5$ Hz), 3.86 (t, 1H, H-4), 3.81 (t, 1H, H-6A), 3.71 (td, 1H, H-1ax, $J_{1ax,2ax} = 12.0$ Hz, $J_{1ax,2eq} = 1.5$ Hz), 3.54 (td, 1H, H-5, $J_{5,4} = J_{5,6A} = 10.0$ Hz), 2.35 (ddt, 1H, H-2eq, $J_{2eq,2ax} = 13.0$ Hz), 1.95 (tdd, 1H, H-2ax); ¹³C NMR (CDCl₃, 100 MHz) δ ppm 165.8 (CO), 137.1 and 130.0 (ArC), 132.9, 129.6 (2C), 128.8, 128.2 (2C), 128.0 (2C) and 126.0 (2C) (ArCH), 101.4 (Ph*C*H(O) (O)-), 80.4, 71.8 and 71.5 (C-3 to C-5), 68.8 and 65.9 [(-), C-6 and C-1], 31.5 [(-), C-2]; IR (film, cm⁻¹) 2866, 1719, 1452, 1315, 1270, 1133, 1103, 1014; HR-FABMS calcd for C₂₀H₂₀O₅Na *m*/*z* 363.12084, found 363.12088.

3-O-Benzoyl-6-O-benzyl-1,2-dideoxy-D-glucopyranose (15). Hydrogen chloride in diethyl ether was added at rt to 14 (1.31 g, 3.84 mmol) and sodium cyanoborohydride³⁶ (966 mg, 4 equiv) in THF (87 mL) containing 4 Å molecular sieves until the evolution of gas ceased. The mixture was filtered through a Celite pad, diluted with dichloromethane (200 mL), and washed with water and saturated aqueous hydrogen carbonate. The organic layer was dried (sodium sulfate) and concentrated, and the resulting syrup was purified by flash chromatography on silica gel, eluting with ethyl acetate/hexanes 3:7, to give **15** (1.21 g, 92%) as a clear syrup: $[\alpha]_D$ –28.4 (*c* 1.0, CHCl₃); ¹H NMR (CDCl₃, 400 MHz, assigned by COSY45) δ 8.06 (d, 2H, ArH, J = 7.0 Hz), 7.57-7.29 (3m, 8H, ArH), 5.08 (ddd, 1H, H-3, $J_{3,2ax} = 11.4$ Hz, $J_{3,4} = 9.0$ Hz, $J_{3,2eq} = 5.2$ Hz), 4.61 (AB, 2H, PhC H_2 -, J = 12.0 Hz), 4.05 (ddfd, 1H, H-1eq, $J_{1eq,1ax} = 12.0$ Hz, $J_{1eq,2ax} = 4.9$ Hz, $J_{1eq,2eq} = 1.5$ Hz), 3.78-3.73 (m, 3H, H-4, H-6 and H-6A), 3.55 (td, 1H, H-1ax, J_{1ax,2ax} = 12.0 Hz, $J_{1ax,2eq}$ = 2.0 Hz), 3.45 (dt, 1H, H-5, $J_{5,4}$ = 9.4 Hz, $J_{5,6}=J_{5,6A}=4.0$ Hz), 2.14 (ddt, 1H, H-2eq, $J_{2\rm eq,2ax}=12.0$ Hz), 1.86 (tdd, 1H, H-2ax); $^{13}{\rm C}$ NMR (CDCl₃, 100 MHz) δ 166.7 (CO), 137.7 and 129.7 (ArC), 133.1, 129.6 (2C), 128.2 (3C), 127.7 (2C) and 127.6 (2C) (ArCH), 79.0, 76.0 and 70.9 (C-3 to C-5), 73.6, 70.1 and 65.3 [(–), Ph CH_2- , C-6 and C-1], 30.8 [(–), C-2]; IR (film, cm $^{-1}$) 3471, 2862, 1718, 1452, 1317, 1273, 1122, 1092, 1071, 1027; HR-FABMS calcd for C₂₀H₂₂O₅Na m/z 365.13649, found 365.13638.

Di(S-2-pyridyl) Thiocarbonate. To a cold solution (0 °C) of 2-mercaptopyridine (8.89 g, 80 mmol) and triphosgene (3.95 g, 13.3 mmol) in dichloromethane (400 mL) was added dropwise triethylamine (12 mL, 86 mmol); the mixture was stirred at this temperature for 30 min and then at rt for 1 h. The mixture was concentrated, treated with cold saturated aqueous bicarbonate, and extracted with 400 mL of ethyl acetate. After being washed with water and brine, the organic layer was dried (sodium sulfate), filtered, and concentrated to give the product as a yellow solid that was dried in vacuo overnight. Pale yellow needle-shaped crystals were obtained by recrystallization from 2-propanol (8.32 g, 84%): mp 44 °C; ¹H NMR (CDCl₃, 400 MHz) δ 8.57 (ddd, 2H, J = 4.8, 1.9, 0.8 Hz), 7.69 (td, 2H, J = 7.9, 1.9 Hz), 7.63 (dt, 2H, J = 7.9, 1.1 Hz), 7.26 (ddd, 2H, J = 7.3, 4.8, 1.3 Hz); ¹³C NMR (CDCl₃, 100 MHz) δ 185.6 (CO), 150.6 (PyC), 150.5, 137.5, 130.5 and 124.2 (PyCH); IR (film, cm⁻¹) 3047, 1712, 1656, 1572, 1562, 1449, 1421, 1282, 1152, 1113, 1083, 1045; HR-FABMS calcd for C11H9ON2S2 m/z 249.01563, found 249.01553.

2,3,4-Tri-*O***-benzyl**- β -L-**fucopyranosyl 2-Thiopyridyl Carbonate (16).** A mixture of 2,3,4-tri-*O*-benzyl-L-fucopyranose⁴⁰ (714 mg, 1.65 mmol), di(*S*-2-pyridyl) thiocarbonate (1.22 g, 3 equiv), and triethylamine (690 μ L, 3 equiv) in dichloromethane (16.5 mL) was stirred at rt for 24 h. Concentration and purification by flash chromatography on silica gel, eluting with ethyl acetate/hexanes 3:7, gave **16** (840 mg, 89%) as a yellow powder: [α]_D -9.3 (*c* 1.0, CHCl₃); ¹H NMR (CDCl₃, 400 MHz, assigned by COSY45) δ 8.60 (d, 1H, PyH, *J* = 5.0 Hz), 7.21 (m, 2H, ArH), 7.41–7.25 (m, 16H, ArH), 5.66 (d, 1H, H-1, *J*_{1,2} = 8.0 Hz), 5.00 (d, 1H, PhC*H*H–, *J* = 11.6 Hz), 4.80 (s, 2H, PhC*H*₂–), 4.78 (AB, 2H, PhC*H*₂–, *J* = 11.9 Hz), 4.72 (d, 1H, PhCH*H*–, *J* = 11.6 Hz), 4.00 (t, 1H, H-2, *J*_{2,3} = 8.0 Hz), 3.68–3.58 (m, 3H, H-3, H-4 and H-5), 1.22 (d, 3H, -CH₃, *J*_{Me,5} = 6.4

⁽⁴⁰⁾ Dejter-Juszynski, M.; Flowers, H. M. Carbohydr. Res. 1971, 18, 219.

Hz); ¹³C NMR (CDCl₃, 100 MHz) δ 167.7 (CO), 151.4 (PyC), 149.9, 137.1, 129.1, 128.3, 128.2, 128.1, 127.6, 127.4 and 123.4 (PhCH and PyCH), 138.1, 138.0, 137.9 (PhC), 96.9 (C-1), 82.3, 77.5, 75.7 and 71.6 (C-2 to C-5), 75.2, 74.6 and 73.0 [(-), PhCH₂-], 16.5 (-CH₃); IR (film, cm⁻¹) 2874, 1736, 1573, 1497, 1454, 1422, 1102, 1062, 1021; HR-FABMS calcd for C₃₂H₃₃O₄-NSNa *m*/*z* 550.20280, found 550.20292.

(2,3,4-Tri-*O*-benzyl- α -L-fucopyranosyl)-*O*-(1 \rightarrow 4)-3-*O*benzoyl-6-O-benzyl-1,2-dideoxy-D-glucopyranose (17). A mixture of 15 (343 mg, 1.00 mmol), 16 (800 mg, 1.4 equiv), 1,1,3,3-tetramethylurea (170 μ L, 1.4 equiv), and activated 4 Å molecular sieves in dichloromethane (38.5 mL) was stirred overnight at rt and then cooled at 0 °C. Silver triflate (2.32 g, 9 equiv) was added to the reaction mixture, and the stirring was continued 24 h at rt in the dark. The suspension was treated with a few drops of pyridine, filtered through Celite, and concentrated. Purification by flash chromatography on silica gel, eluting with ethyl acetate/hexanes 2:8, gave 17 (758 mg, > 98%) as a syrup: $[\alpha]_D$ –48.6 (c 1.0, CHCl₃); ¹H NMR (CDCl₃, 400 MHz, assigned by COSY45) δ 8.08 (d, 2H, ArH, J = 7.0 Hz), 7.65–7.20 (3m, 23H, ArH), 5.30 (ddd, 1H, H-3, $J_{3,2ax}$ = 11.0 Hz, $J_{3,4}$ = 9.0 Hz, $J_{3,2eq}$ = 5.5 Hz), 5.07 (d, 1H, H-1', $J_{1',2'} = 3.5$ Hz), 4.92 (d, 1H, PhC*H*H-, J = 11.5 Hz), 4.84 (d, 1H, PhC*H*H-, J = 11.6 Hz), 4.74 (AB, 2H, PhC H_2 -, J = 11.7 Hz), 4.62 (d, 1H, PhCH*H*-, J = 11.6 Hz), 4.58 (d, 1H, PhCHH-, J = 11.5 Hz), 4.46 (s, 2H, PhCH₂-), 4.06 (dd, 1H, H-1eq, $J_{1eq,1ax} = 11.6$ Hz, $J_{1eq,2ax} = 4.0$ Hz, $J_{1eq,2eq} = 0$ Hz), 4.00 (dd, 1H, H-2', $J_{2',3'} = 10.3$ Hz), 3.95–3.84 (m, 5H, H-4, H-6, H-6A, H-3' and H-5'), 3.60-3.51 (m, 3H, H-1ax, H-5 and H-4'), 2.20 (brdd, 1H, H-2eq, J_{2eq,2ax} = 11.0 Hz), 1.86 (brq, 1H, H-2ax, $J_{2ax,1ax} = 11$ Hz), 0.78 (d, 3H, $-CH_3$, $J_{Me,5'} = 6.4$ Hz); ¹³C NMR $(CDCl_3, 100 \text{ MHz}) \delta 166.2 (CO), 138.9, 138.7 (2C), 138.4 \text{ and}$ 130.5 (ArC), 133.1, 129.9, 128.4,128.3, 128.2, 128.0, 127.8, 127.7, 127.6 and 127.5 (ArCH), 99.4 (C-1'), 79.9, 79.5, 77.8, 76.6, 76.5, 74.6 and 67.2 (C-3 to C-5 and C-2' to C-5'), 75.0, 74.2, 73.5, 72.9, 69.6, and 65.4 [(-), PhCH₂-, C-6 and C-1], 31.6 [(-), C-2], 16.4 (-CH₃); IR (film, cm⁻¹) 3031, 2864, 1717, 1454, 1273, 1099, 1069, 1047, 1028; HR-FABMS calcd for C47H50O9Na m/z 781.33525, found 781.33512.

(2,3,4-Tri-O-benzyl- α -L-fucopyranosyl)-O-(1 \rightarrow 4)-6-Obenzyl-1,2-dideoxy-D-glucopyranose (18). To a solution of 17 (758 mg, 1.00 mmol) in methanol (5 mL) was added 2 mL (1.0 equiv) of a freshly prepared methanol solution of sodium methoxide (0.5 M). The solution was stirred at 45 °C for 2 h and neutralized with Amberlite IR-120 (H⁺). After filtration and concentration, the resulting syrup was purified by flash chromatography on silica gel, eluting with ethyl acetate/ hexanes 3:7, to give **18** (514 mg, 85%) as a white powder: $[\alpha]_D$ -24.2 (c 1.1, CHCl₃); ¹H NMR (CDCl₃, 400 MHz, assigned by COSY45) δ 7.45–7.20 (m, 20H, ArH), 5.04 (d, 1H, PhCHH–, J = 11.4 Hz), 5.00 (d, 1H, H-1', $J_{1',2'} = 3.8$ Hz), 4.88 (d, 1H, PhC*H*H-, *J* = 11.9 Hz), 4.81 (d, 1H, PhC*H*H-, *J* = 11.9 Hz), 4.78 (d, 1H, PhCHH-, J = 12.1 Hz), 4.70 (d, 1H, PhCHH-, J = 11.4 Hz), 4.68 (d, 1H, PhCHH-, J = 11.7 Hz), 4.62 (brs, 1H, -OH), 4.42 (AB, 2H, PhCH₂-, J = 12.3 Hz), 4.15-4.07 (m, 2H, H-2' and H-5'), 4.04 (dd, 1H, H-1eq, $J_{1eq,1ax} = 11.5$ Hz, $J_{1eq,2ax} = 4.0$ Hz), 3.99 (dd, 1H, H-6, $J_{6,6A} = 10.5$ Hz, $J_{6,5} = 1.6$ Hz), 3.93 (dd, 1H, H-3', $J_{3',2'} = 10.2$ Hz, $J_{3',4'} = 2.6$ Hz), 3.74 (fd, 1H, H-4'), 3.73 (dd, 1H, H-6A, $J_{6A,5} = 6.3$ Hz), 3.63 (m, 1H, H-3), 3.50-3.40 (m, 2H, H-1ax and H-5), 3.28 (t, 1H, H-4, $J_{4,3} = J_{4,5} = 8.6$ Hz), 2.03 (brdd, 1H, H-2eq, $J_{2eq,2ax} = 11.0$ Hz, $J_{2eq,3} = 4.0$ Hz), 1.73 (brq, 1H, H-2ax, $J_{2ax,1ax} = J_{2ax,3} = 11$ Hz, $J_{2ax,1eq} = 4$ Hz), 1.22 (d, 3H, $-CH_3$, $J_{Me,5'} = 6.4$ Hz); ¹³C NMR (CDCl₃, 100 MHz) δ 138.7, 138.6, 138.4 (2C) (ArC), 128.5, 128.4, 128.3, 128.0, 127.9, 127.7, 127.6, 127.5 and 127.4 (ArCH), 99.7 (C-1'), 84.0, 79.0, 78.3, 77.4, 76.0, 71.9 and 68.0 (C-3 to C-5 and C-2' to C-5'), 75.0, 73.8, 73.3, 73.2, 70.1, and 65.7 [(-), PhCH₂-, C-6 and C-1], 32.8 [(-), C-2], 16.8 (-CH₃); HR-FABMS calcd for C40H46O8Na m/z 677.30904, found 677.30865.

(2-*O*-Benzoyl-4,6-*O*-benzylidene-3-*O*-[(*S*)-1-(methoxycarbonyl)-2-cyclohexyl-ethyloxy]- β -D-galactopyranosyl)-*O*-(1 \rightarrow 3)-[(2,3,4-tri-*O*-benzyl- α -L-fucopyranosyl)-*O*-(1 \rightarrow 4)]-6-*O*-benzyl-1,2-dideoxy-D-glucopyranose (19). To a cold solution of 18 (96 mg, 146.8 μ mol) in THF (14.7 mL) was added sodium hydride (9 mg, 1.5 equiv), and the mixture was stirred at 0 °C for 30 min. Silver triflate (42 mg, 1.1 equiv) was added, the mixture stirred in the dark 15 min, 11 (66.3 mg, 0.7 equiv) and 4 Å molecular sieves were then added, and the mixture was allowed to warm to rt in the dark overnight. The mixture was filtered through Celite, concentrated, redissolved in dichloromethane (20 mL), and washed with saturated aqueous ammonium cloride (10 mL) and brine (10 mL). The organic phase was dried (sodium sulfate), concentrated, and purified by flash chromatography on silica gel, eluting with ethyl acetate/hexanes 6:4 gave 19 (55.3 mg, 48%) as a syrup: $[\alpha]_D$ -31.0 (c 0.6, CHCl₃); ¹H NMR (CDCl₃, 400 MHz, assigned by COSY45) δ 8.07 (dfd, 2H, ArH, J = 8.3, 1.2 Hz), 7.68 (d, 2H, ArH, J = 7.0 Hz), 7.58 (t, 1H, ArH, J = 7.0 Hz), 7.46 (t, 2H, ArH, J = 7.0 Hz), 7.40–7.10 (m, 23H, ArH), 5.67 (s, 1H, PhCH(O)(O)-), 5.62 (t, 1H, H-2', $J_{2',1'} = J_{2',3'} = 8.8$ Hz), 4.91 (q, 1H, H-5", $J_{5",Me} = 6.4$ Hz), 4.85 (d, 1H, H-1", $J_{1",2"} = 3.3$ Hz), 4.75 (d, 1H, PhC*H*H-, J = 11.7 Hz), 4.66 (d, 1H, PhC*H*H-, J = 11.5 Hz), 4.65 (d, 1H, H-1', $J_{1',2'} = 7.0$ Hz), 4.57 (d, 1H, PhCHH-, J = 11.5 Hz), 4.55 (d, 1H, PhCHH-, J =11.7 Hz), 4.49 (fd, 1H, H-4', $J_{4,3} = 2.9$ Hz), 4.34 (s, 2H, PhC- H_2 -), 4.33 (d, 1H, H-6', $J_{6',6A'}$ = 12.6 Hz), 4.22 (d, 1H, PhCHH-J = 11.3 Hz), 4.18–4.13 (m, 2H, $-OCHCO_2Me$ and H-6A'), 3.97-3.88 (m, 2H, H-2" and H-3"), 3.86-3.56 (m, 10H, H-1eq, H-3, H-4, H-3', H-6, H-6A, PhCHH- and -OCH₃), 3.46 (s,1H, H-5'), 3.31-3.24 (m, 3H, H-1ax, H-5 and H-4"), 1.83 (brdd, 1H, H-2eq, $J_{2eq,2ax} = 12.0$ Hz, $J_{2eq,3} = 5.0$ Hz), 1.67 (m, 1H, -CHHCHCO2Me-), 1.52-1.27 (m, 8H, H-2ax, -CH2- and -CHHCHCO₂Me-), 1.26 (d, 3H, -CH₃), 0.96-0.71 (m, 5H, $-CH_2-$ and -CH-); ¹³C NMR (CDCl₃, 100 MHz) δ 174.3 (CO_2- Me), 164.8 (PhCO), 140, 139.6, 138.9, 138.3 and 130.1 (ArC), 133.3, 129.9, 128.9, 128.8, 128.6, 128.4, 128.3, 128.2, 128.0, 127.9, 127.8, 127.7, 127.6, 127.3, 127.0 and 126.2 (ArCH and ArC), 99.8, 99.3 and 97.6 (PhCH(O)(O)-, C-1' and C-1"), 80.4, 80.1, 78.9 (2C), 78.5, 78.3, 75.6, 75.1, 72.5, 71.6, 66.7 and 65.8 (C-3 to C-5, C-2' to C-5', C-2" to C-5" and -OCHCO2Me), 75.0, 74.7, 73.4, 71.3, 69.4, 68.4 and 66.6 [(-), PhCH₂-, C-1, C-6 and C-6'], 52.1 (-OCH₃), 41.0, 33.9, 32.6, 31.4, 26.4, 26.0 and 25.7 [(-), -CH₂- and C-2], 33.4 (-CH-), 16.3 (-CH₃); HR-FABMS calcd for $C_{70}H_{80}O_{16}Na$ m/z 1199.53440, found 1199.53391

(2-O-Benzoyl-4,6-O-benzylidene-3-O-[(S)-1-(oxycarbonyl)-2-cyclohexylethyloxy]- β -D-galactopyranosyl)-O-(1 -3)-[(2,3,4-tri-O-benzyl- α -L-fucopyranosyl)-O-(1 \rightarrow 4)]-6-Obenzyl-1,2-dideoxy-D-glucopyranose (20). To a cooled solution (0 °C) of 19 (94.2 mg, 80.06 µmol) in THF (8 mL) was added lithium hydroxide (3.7 mg, 1.1 equiv) dissolved in water (470 μ L). The mixture was stirred at rt for 48 h and neutralized with Amberlite IR-120 (H⁺). After filtration and concentration, the residue was purified by flash chromatography on silica gel, eluting with dichloromethane/methanol 14:1, to give 20 (93 mg, >98%) as a gum: $[\alpha]_D$ -86.0 (*c* 0.2, CHCl₃); ¹H NMR (CD₃-OD, 400 MHz) δ ppm 8.16 (d, 2H, ArH, J = 8.0 Hz), 7.74 (d, 2H, ArH, J = 7.7 Hz), 7.65 (t, 1H, ArH, J = 7.0 Hz), 7.53 (t, 2H, ArH, J = 7.0 Hz), 7.40-7.19 (m, 23H, ArH), 5.75 (s, 1H, PhC*H*(O)(O)-), 5.61 (t, 1H, H-2', $J_{2',1'} = J_{2',3'} = 9.0$ Hz), 5.06 (q, 1H, H-5", $J_{5",Me} = 6.4$ Hz), 4.83 (d, 1H, H-1", $J_{1",2"} = 3.8$ Hz), 4.79 (d, 1H, H-1', $J_{I,Z} = 8.0$ Hz), 4.72 (d, 1H, J = 11.9Hz), 4.67 (d, 1H, J = 11.4 Hz), 4.61 (d, 1H, J = 11.6 Hz), 4.59 (brs, 1H, H-4'), 4.42 (t, 2H, J = 11.4 Hz), 4.27-4.18 (m, 5H), 3.96-3.91 (m, 2H), 3.79-3.73 (m, 4H), 3.66-3.49 (m, 5H), 3.40 (brs, 1H), 3.20 (brd, 1H, J = 9 Hz), 2.00 (brdd, 1H, H-2eq, $J_{2eq,2ax} = 12.0$ Hz, $J_{2eq,3} = 4.0$ Hz), 1.70–0.80 (2m, 14H, H-2ax, -CH₂- and -CH-), 1.26 (d, 3H, -CH₃); ¹³C NMR (CDCl₃, 100 MHz) & 178.9 (CO₂H), 166.7 (PhCO), 139.9, 139.6, 139.0, 138.3 (2C) and 129.9 (ArC), 134.0, 130.3, 129.2, 128.9, 128.6, 128.3, 128.1, 128.0, 127.9, 127.8, 127.6, 127.4, and 126.3 (ArCH), 100.1, 99.3 and 97.7 (PhCH(O)(O)-, C-1' and C-1"), 80.9, 80.7, 80.4, 79.0 (2C), 75.9, 75.8, 74.1, 72.7, 71.1, 66.8 and 66.7 (C-3 to C-5, C-2' to C-5', C-2" to C-5" and -OCHCO₂H), 75.4, 74.9, 73.6, 71.7, 69.5, 68.6 and 66.0 [(-), PhCH₂-, C-1, C-6 and C-6'], 41.7, 33.5, 31.8, 30.1, 26.7, 26.2 and 26.1 [(-), -CH₂- and C-2], 34.0 (-CH-), 16.7 (-CH₃); IR (film, cm⁻¹) 2924, 1732, 1603, 1452, 1365, 1268, 1097, 1057; HR-FABMS calcd for C₆₉H₇₈O₁₆-Na m/z 1185.51880, found 1185.52240.

(4,6-O-Benzylidene-3-O-[(S)-1-(oxycarbonyl)-2-cyclohexylethyloxy]- β -D-galactopyranosyl)-O-(1 \rightarrow 3)-[(2,3,4tri-*O*-benzyl- α -L-fucopyranosyl)-*O*-(1 \rightarrow 4)]-6-*O*-benzyl-1,2-dideoxy-D-glucopyranose (21). To a solution of 20 (44.1 mg, 37.93 μ mol) in methanol (5.4 mL) was added 1.52 mL (20 equiv) of a freshly prepared methanol solution of sodium methoxide (0.5 M). The solution was refluxed for 2 h and neutralized with Amberlite IR-120 (H⁺). After filtration and concentration, the resulting syrup was purified by flash chromatography on silica gel, eluting with dichloromethane/ methanol 12:1, to give **21** (36 mg, 90%) as a gum: $[\alpha]_D - 107.2$ (c 0.18, CH₃OH); ¹H NMR (CD₃OD, 400 MHz, assigned by COSY45) δ 7.66 (d, 2H, ArH, J = 7.4 Hz), 7.40–7.10 (m, 23H, ArH), 5.68 (s, 1H, PhC*H*(O)(O)–), 5.01 (q, 1H, H-5", $J_{5",Me} = 6.4$ Hz), 4.89 (d, 1H, H-1", $J_{1",2"} = 3.8$ Hz), 4.72–4.67 (m, 3H, $-OCHCO_2Me$ and 2 PhCHH-), 4.58 (d, 1H, PhCHH-, J = 11.9 Hz), 4.47 (d, 1H, PhCHH-, J = 11.8 Hz), 4.45 (d, 1H, PhCH*H*-, J = 11.4 Hz), 4.42 (d, 1H, H-4', $J_{4',3'} = 3.3$ Hz), 4.36 (d, 1H, H-1', $J_{1',2'} = 7.7$ Hz), 4.22–4.15 (m, 3H, H-6', H-6A' and PhCHH-), 3.95-3.78 (m, 6H, H-1eq, H-3, H-6, H-2', H-2" and H-3"), 3.68 (t, 1H, H-4, $J_{4,3} = J_{4,5} = 9.4$ Hz), 3.60-3.47 (m, 4H, H-6A, H-3', H-5' and PhCHH-), 3.39 (brt, 1H, H-1ax, $J_{1ax,2ax} = J_{1ax,1eq} = 9.0$ Hz), 3.31 (1H, H-4''), 3.24 (m, 1H, H-5), 2.10 (brdd, 1H, H-2eq, $J_{2eq,2ax} = 12.0$ Hz, $J_{2eq,3} = 5.0$ Hz), 2.00 (brd, 1H, J = 11.0 Hz), 1.90–1.55 (m,6H), 1.45–1.20 (m,5H) and 1.03-0.94 (m, 2H) (-CH2-, H-2ax and -CH-), 1.09 (d, 3H, -CH₃); ¹³C NMR (CD₃OD, 100 MHz) δ 178.3 (CO), 140.5, 140.4, 140.2, 139.7 and 139.3 (ArC), 129.6, 129.4, 129.3, 129.1, 129.0, 128.7, 128.6, 128.5, 128.4, 128.2 and 127.2 (ArCH), 102.1, 100.8 and 98.7 (PhCH(O)(O)-, C-1' and C-1"), 81.3, 80.4, 80.0 (2C), 78.5, 78.4, 76.9, 76.6, 74.6, 72.0, 67.9 and 67.7 (C-3 to C-5, C-2' to C-5', C-2" to C-5" and -OCHCO2H), 76.3, 75.4, 74.4, 72.4, 70.7, 69.5 and 66.8 [(-), PhCH2-, C-1, C-6 and C-6'], 41.9, 35.1, 33.8, 32.5, 27.7, 27.6 and 27.4 [(-), -CH₂- and C-2], 35.4 (–CH–), 16.8 (–CH₃); HR-FABMS calcd for $C_{62}H_{73}O_{15}$ m/z 1057.49500, found 1057.49130.

(3-*O*-[(*S*)-1-(Oxycarbonyl)-2-cyclohexylethyloxy]-β-D-galactopyranosyl)-*O*-(1 → 3)-[α-L-fucopyranosyl-*O*-(1 → 4)]-1,2-dideoxy-D-glucopyranose Sodium Salt (2). To a solution of 21 (25 mg, 23.6 µmol) dissolved in a mixture of dioxane (5.50 mL) and water (2.40 mL) was added 20% palladium hydroxide on carbon (Degussa, 18.0 mg) and acetic acid (24 µL, 18 equiv). The resulting mixture was hydrogenated under 60 psi at rt for 40 h. After filtration on Celite and

concentration, the product was passed through an ionexchange resin column (Dowex in the sodium form); the solution afforded after rinsing the column with water and lyophilizing the eluate pure 2 as a sodium salt (14.4 mg, >98%) as a white powder: $[\alpha]_D - 49.0$ (*c* 1.06, H₂O); ¹H NMR (D₂O CH₃OD as external standard δ = 3.35 ppm, 400 MHz, assigned by COSY45) δ 4.97 (d, 1H, H-1", $J_{1",2"} = 3.9$ Hz), 4.77 (q, 1H, H-5"), 4.52 (d, 1H, H-1', $J_{1',2'} = 7.9$ Hz), 4.10–3.92 (m, 3H, H-3, H-1eq and-OCHCO2Me), 3.92 (fd, 1H, H-4', J_{4',3'} = 3.0 Hz), 3.90-3.85 (m, 3H, H-6, H-6A and H-3"), 3.82 (fd, 1H, H-4" $J_{4'',3''} = 3.0$ Hz), 3.78 (dd, 1H, H-2", $J_{2'',3''} = 10.5$ Hz), 3.74 (brd, 2H, H-6' and H-6A', J = 6 Hz), 3.64 (t, 1H, H-2', $J_{2',3'} = 8.0$ Hz), 3.62 (t, 1H, H-5', $J_{5',6'} = J_{5',6A'} = 6.0$ Hz), 3.59 (t, 1H, H-4, $J_{4,3} = J_{4,5} = 9.0$ Hz), 3.50 (brt, 1H, H-1ax, $J_{1ax,2ax} = J_{1ax,1eq} = 11.5$ Hz), 3.42–3.39 (m, 2H, H-5 and H-3'), 2.23 (brdd, 1H, H-2eq, J_{2eq,2ax} = 12.0 Hz, J_{2eq,3} = 4.0 Hz), 1.80 (m, 1H), 1.75-1.50 (m, 8H), 1.35-1.10 (m, 6H) and 1.00-0.85 (m, 2H) (-CH₂-, H-2ax and -CH-), 1.21 (d, 3H, -CH₃, J_{Me,5"} = 6.5 Hz); ¹³C NMR (D₂O, CH₃OD as external standard δ = 49.6 ppm, 100 MHz) δ 183.1 (CO), 99.4 and 99.0 (C-1' and C-1"), 83.1, 80.4, 79.6, 76.0, 75.0, 74.6, 72.2, 70.1, 69.6, 68.3, 67.2 and 66.6 (C-3 to C-5, C-2' to C-5', C-2" to C-5" and -OCHCO2H), 65.5, 61.9 and 60.3 [(-), C-1, C-6 and C-6'], 41.5, 33.9, 32.2, 30.6, 26.5, 26.3, and 26.0 [(-), -CH₂- and C-2], 33.6 (-CH-), 15.9 (-CH₃); IR (KBr, cm⁻¹) 3430, 2925, 1599, 1400, 1079; HR-FABMS calcd for $C_{27}H_{46}O_{15}Na m/z 633.27344$, found 633.27160.

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Supporting Information Available: Copies of selected ¹H and ¹³C spectra and X-ray structure determination data. This material is available free of charge via the Internet at http://pubs.acs.org.

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